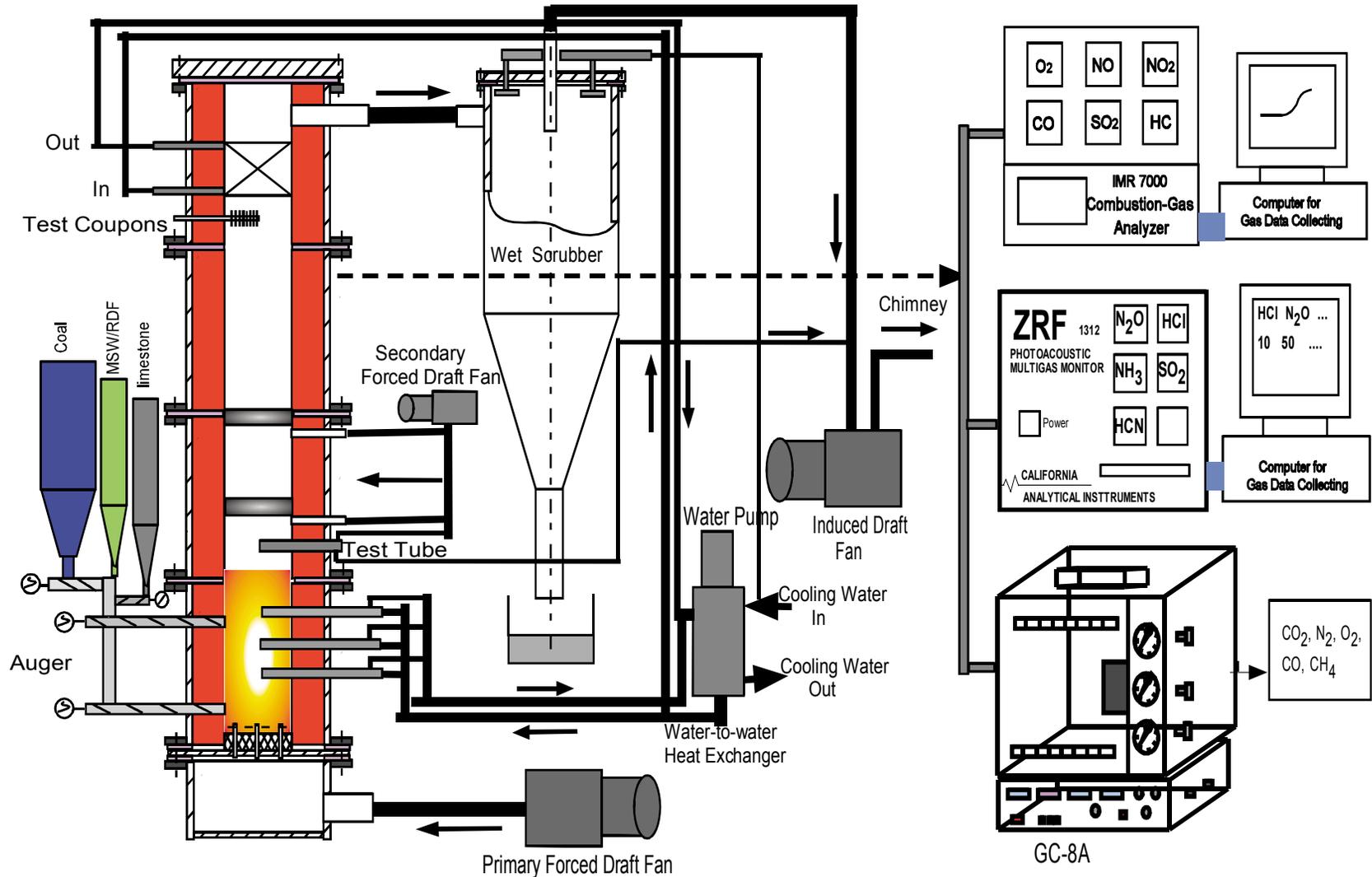


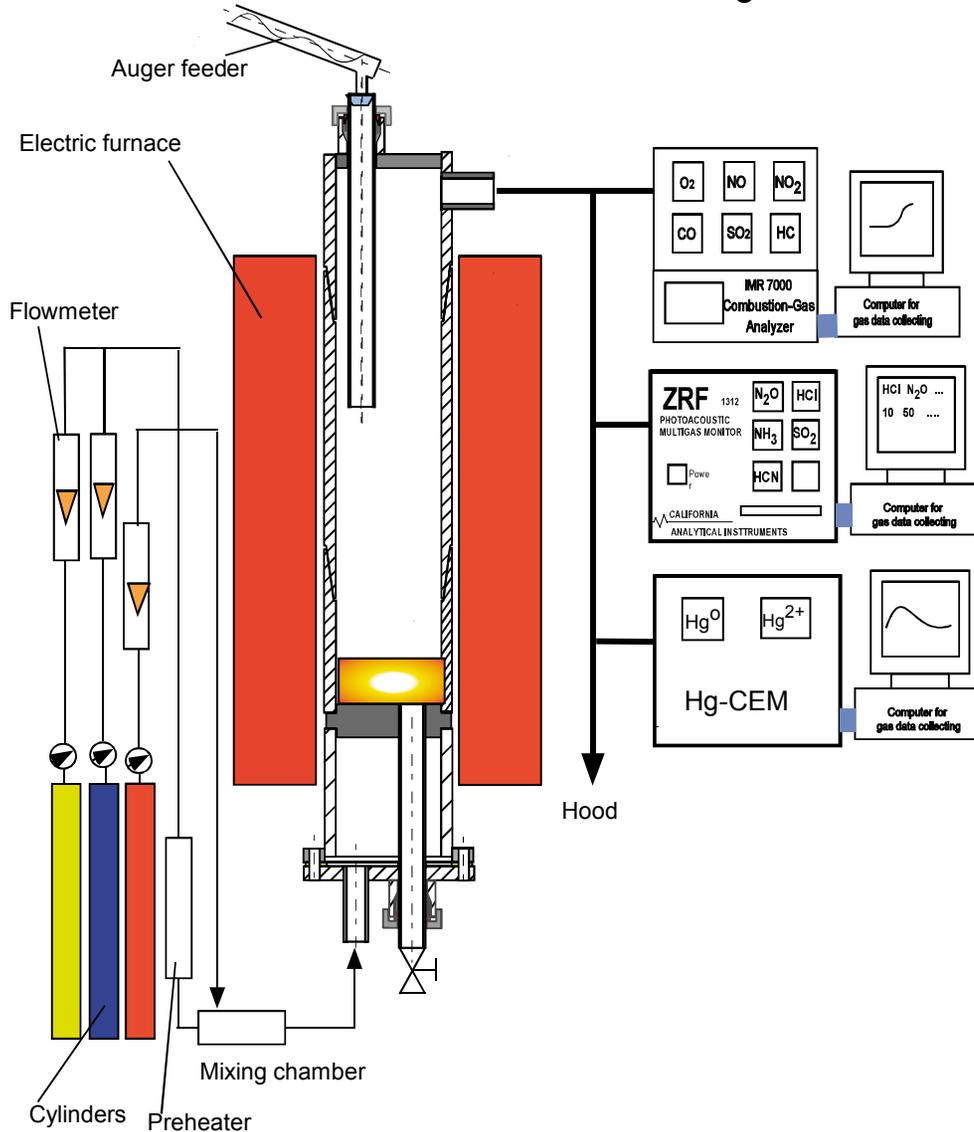
**ESTABLISHMENT OF AN  
ENVIRONMENTAL CONTROL  
TECHNOLOGY LABORATORY WITH A  
CIRCULATING FLUIDIZED BED  
COMBUSTION SYSTEM**

**Materials Characterization Center  
Western Kentucky University  
Bowling Green, KY 42101**

# Schematic Diagram of WKU Combustion FBC System



# Schematic Diagram of Drop-Tube Furnace System



# Main Parameters of Combustors

- **Combustor Size:**
  - 12” I.D. and 24” O.D. with 15 ft height
- **Capacity:**
  - 20-30 lb/h coal, 4 lb/min air
- **Sample Location:**
  - Alloy specimen -- 10 ft above distributor
  - A210-C -- 5.3 ft above distributor
- **Identified Components in Flue Gas:**
  - SO<sub>2</sub>, HCl, O<sub>2</sub>, CO<sub>2</sub>, H<sub>2</sub>O and CO

# Identified Components in Flue Gas

- Inorganic:
  - $\text{SO}_2$ , HCl, nitrogen compounds ( $\text{NO}$ ,  $\text{NO}_2$ ,  $\text{N}_2\text{O}$ , HCN,  $\text{NH}_3$ )
- Organic:
  - PAHs (polycyclic aromatic hydrocarbons)
  - PCDD/Fs (polychlorinated dibenzo-p-dioxins/dibenzofurans)

# Recent Research Projects Conducted with the 0.1 MW<sub>th</sub> FBC

- Emission Studies:
  - Behavior of Nitrogen Compounds in NO<sub>x</sub> Selective Catalytic Reduction (SCR) Systems, Tennessee Valley Authority. Project No. 98PPW-242520
  - Co-Firing High Sulfur Coal with Refuse Derived Fuels. DOE Project No. DE-FG-94PC-94211
- Corrosion Research:
  - A Study of Chlorine in High Temperature Corrosion of Alloys in an AFBC System. Project No. ICCI-97-06
  - Behavior of Chlorine during Coal Combustion in AFBC System. EPRI Project No. WO9002-13

# Analytical Values<sup>A</sup> for the Coals Used in the Study

	Phase 0 (1995-1997)		Phase I (1998-1999)			Phase II (1999)	
	<u>95011</u>	<u>95031</u>	<u>97025</u>	<u>98011</u>	<u>98111</u>	<u>99426</u>	<u>99626</u>
<b><u>Proximate Analysis</u></b>							
% Moisture	10.07	8.32	4.56	10.63	10.25	2.33	6.96
% Ash	9.37	10.78	10.97	10.10	9.67	8.70	13.81
% Volatile Matter	43.34	37.21	36.25	34.03	34.84	36.98	34.14
% Fixed Carbon	47.29	52.02	52.78	55.87	55.49	54.32	52.05
<b><u>Ultimate Analysis</u></b>							
<b>% Ash</b>	<b>9.37</b>	<b>10.78</b>	<b>10.79</b>	<b>10.10</b>	<b>9.67</b>	<b>8.70</b>	<b>13.81</b>
% Carbon	74.08	72.16	74.69	75.59	74.18	76.25	69.53
% Hydrogen	5.08	4.82	4.95	4.52	4.73	4.89	4.49
% Nitrogen	1.54	1.54	1.63	1.51	1.59	1.79	1.55
<b>% Sulfur</b>	<b>3.2</b>	<b>2.38</b>	<b>3.06</b>	<b>1.09</b>	<b>1.68</b>	<b>0.97</b>	<b>4.48</b>
% Oxygen	6.72	7.57	4.50	6.73	7.72	7.37	5.73
<b><u>Miscellaneous Analysis</u></b>							
<b>Chlorine (%)</b>	<b>0.012</b>	<b>0.31</b>	<b>0.21</b>	<b>0.47</b>	<b>0.42</b>	<b>0.026</b>	<b>0.41</b>
BTU/pound	13203	12842	13152	13017	13226	13655	12406

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<sup>A</sup> Moisture is as-received, all other values are reported on a dry basis

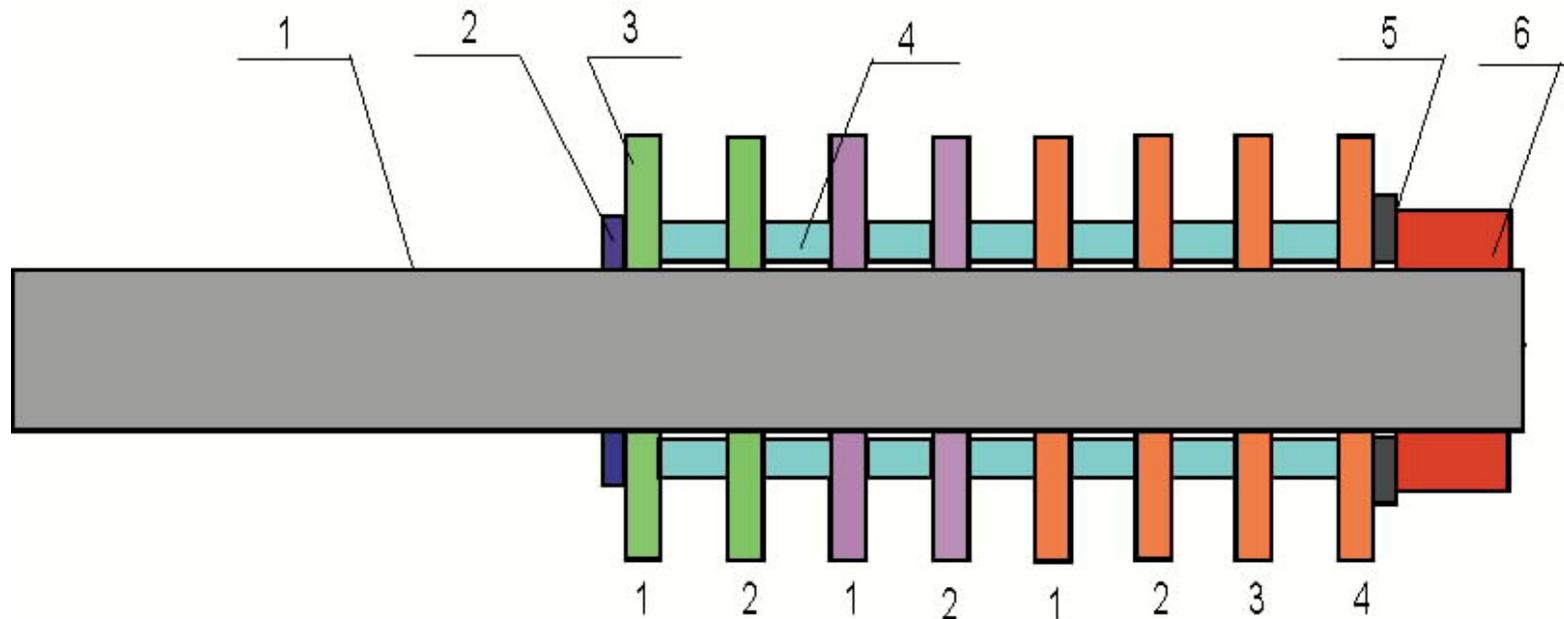
# Analysis of Ashes Prepared from the Coals Used in the Study

<u>Metal Oxides</u>	<u>95011</u>	<u>95031</u>	<u>97025</u>	<u>98011</u>	<u>98111</u>	<u>99426</u>	<u>99626</u>
SiO <sub>2</sub>	49.22	47.32	38.51	54.66	49.03	52.47	31.36
P <sub>2</sub> O <sub>5</sub>	3.36	4.46	0.785	1.87	4.49	2.52	3.21
CaO	1.86	1.62	3.75	1.96	2.63	3.27	6.13
<b>K<sub>2</sub>O</b>	<b>4.14</b>	<b>5.03</b>	<b>1.48</b>	<b>2.91</b>	<b>2.87</b>	<b>1.92</b>	<b>1.39</b>
TiO <sub>2</sub>	1.11	1.14	1.00	1.23	1.17	1.71	0.67
Fe <sub>2</sub> O <sub>3</sub>	19.90	17.04	26.98	9.01	15.44	5.35	36.41
<b>Na<sub>2</sub>O</b>	<b>0.58</b>	<b>1.90</b>	<b>0.41</b>	<b>1.60</b>	<b>1.13</b>	<b>0.38</b>	<b>0.98</b>
MgO	0.70	0.80	0.92	1.08	0.65	0.82	1.42
Al <sub>2</sub> O <sub>3</sub>	17.12	18.18	19.52	23.76	23.33	27.90	12.73
MnO <sub>2</sub>	0.02	0.02	0.01	0.03	0.03	0.023	0.06
<b>Total Alkali</b>	<b>4.72</b>	<b>6.93</b>	<b>1.89</b>	<b>4.51</b>	<b>4.00</b>	<b>2.30</b>	<b>2.73</b>

# Corrosion Test Alloys

- Alloy Specimens
  - SS304 (18-20% Cr, 8-10% Ni), SS347 (17-20% Cr, 8-12% Ni), SS309 (22-26% Cr, 12-14% Ni)
- Cooled and Uncooled Tubes
  - Evaporating Tubes in FBC boiler from TVA
  - A210-C

# The Set-up for Test Coupons



1. stainless steel rod, 2. ceramic washer, 3. test coupons, 4. spacer,  
5. ceramic washer, 6. nut



304



347



309

# Test Conditions

- FBC Operating Parameters:
  - Temperature: 850°C
  - Fluidizing Velocity: 1.25 m/s
  - Excess Air Ratio ( $\alpha$ ): 1.25
  - Ca/S Molar Ratio: 3-3.5
  - Bed Height: 0.6 meter (static)
  - Limestone as bed material
- Coupon Conditions:
  - SS304 (18-20% Cr, 8-10% Ni), SS347 (17-20% Cr, 8-12% Ni), SS309 (22-26% Cr, 12-14% Ni)
  - Location: 3.1 meter above distributor
  - Surface and surrounding temperature: 500-550°C
- A210-C Cooled Tube (I.D. 1 $\frac{1}{2}$ " , O.D. 2")
  - Location: 1.6 meters above distributor
  - Surface temperature: 380-400°C (air coolant)
  - Surrounding temperature: 850°C

# Summary of Steady-State Run Conditions

	<u>Runs</u>						
	<u>1<sup>st</sup></u>	<u>2<sup>nd</sup></u>	<u>3<sup>rd</sup></u>	<u>4<sup>th</sup></u>	<u>5<sup>th</sup></u>	<u>6<sup>th</sup></u>	<u>7<sup>th</sup></u>
<u>Fuel feed rate (kg/h)</u>	8.41	10.44	7.50	7.95	8.21	8.16	7.26
<u>Limestone rate (kg/h)</u>	2.21	2.21	1.61	1.02	1.43	2.72	3.22
<u>Ca/(S+Cl) molar ratio</u>	3.10	3.10	2.88	3.15	3.19	9.21	3.05
Temperature (K)							
At 0.56 meter*		1108	1168	1218			
At 1.90 meter		943	1013	1078			
At 3.30 meter		843	923	978			
SO <sub>2</sub> Concentration (ppm)							
Coupon	250	190	245	174	114	53	282
A210-C tube	270	210	260	220	137	68	286
HCl Concentration (ppm)							
Coupon	1.5	38.5	44.3	65.7	55.3	2.8	55.1
A210-C tube	10	46.7	56.8	84.8	68.2	13.0	85.4
CO Concentration (ppm)							
Coupon	140	160	156	214	156	113	173
A210-C tube	155	165	170	222	187	164	244

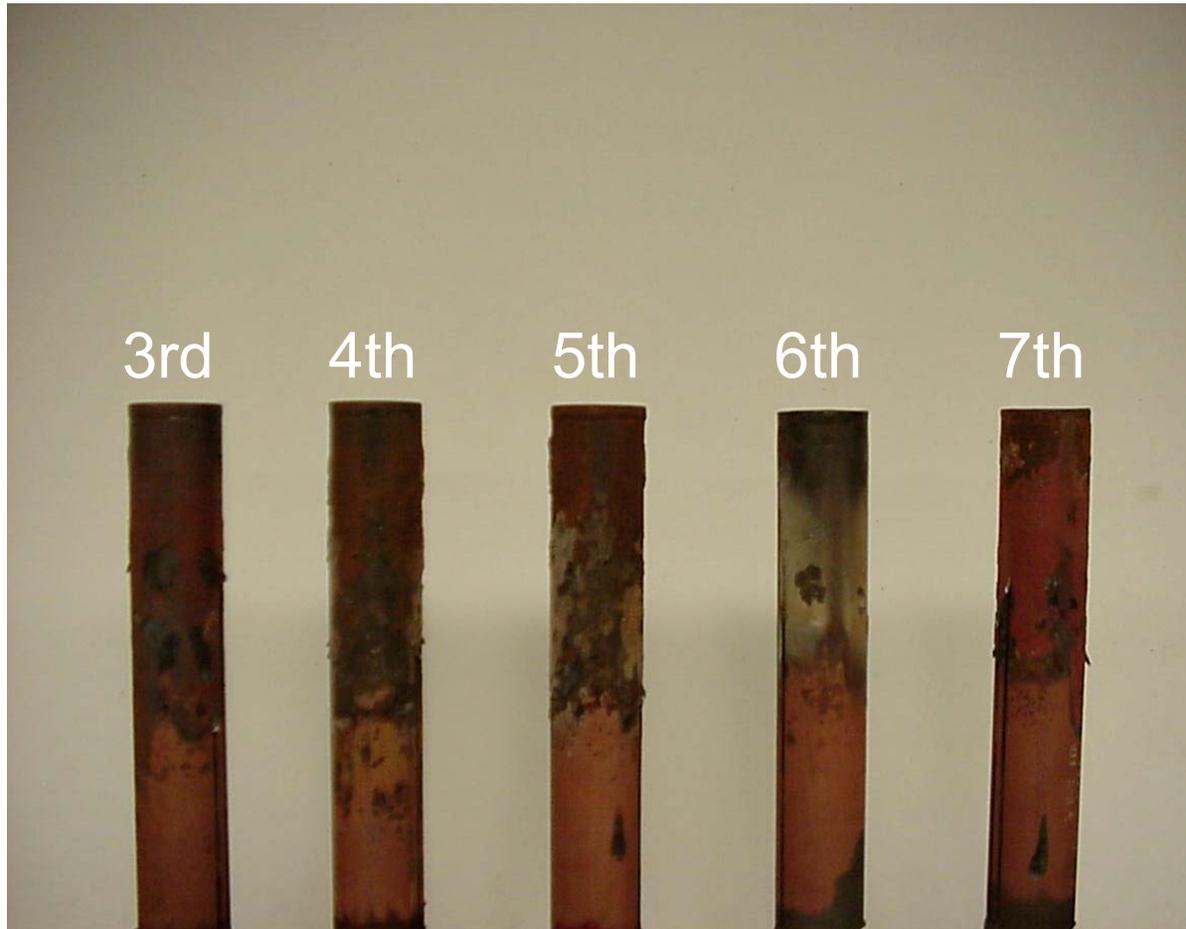
# Final Metal Loss after 1,000 Hours Exposure in 0.1 MWth WKU-FBC System

	Phase 0 (1995-1997)		Phase I (1998-1999)			Phase II (1999)	
	<u>95011</u>	<u>95031</u>	<u>97025</u>	<u>98011</u>	<u>98111</u>	<u>99426</u>	<u>99626</u>
Ultimate Analysis							
<b>% Ash</b>	<b>9.37</b>	<b>10.78</b>	<b>10.79</b>	<b>10.10</b>	<b>9.67</b>	<b>8.70</b>	<b>13.81</b>
<b>% Sulfur</b>	<b>3.2</b>	<b>2.38</b>	<b>3.06</b>	<b>1.09</b>	<b>1.68</b>	<b>0.97</b>	<b>4.48</b>
<b>% Chlorine</b>	<b>0.012</b>	<b>0.31</b>	<b>0.21</b>	<b>0.47</b>	<b>0.42</b>	<b>0.026</b>	<b>0.41</b>
<b>% Alkali</b>	<b>4.72</b>	<b>6.93</b>	<b>1.89</b>	<b>4.51</b>	<b>4.00</b>	<b>2.30</b>	<b>2.73</b>
Thickness of coupon (mils)							
Alloy <b>304</b> : <i>Initial</i>	91.535	103.150	106.142	107.283	108.464	114.370	113.228
<i>After 1,000 hours</i>	90.945	102.480	105.512	106.929	107.874	114.213	112.638
<b>Loss per year* (mils)</b>	<b>2.586</b>	<b>2.931</b>	<b>2.759</b>	<b>1.552</b>	<b>2.586</b>	<b>0.689</b>	<b>2.587</b>
Alloy <b>309</b> : <i>Initial</i>	103.346	134.291	128.150	127.165	101.614	128.583	130.118
<i>After 1,000 hours</i>	103.150	134.095	127.992	127.047	101.457	128.504	129.646
<b>Loss per year(mils)</b>	<b>0.862</b>	<b>0.862</b>	<b>0.670</b>	<b>0.517</b>	<b>0.670</b>	<b>0.345</b>	<b>2.069</b>
Alloy <b>347</b> : <i>Initial</i>	92.244	123.661	115.196	114.213	114.213	119.724	122.441
<i>After 1,000 hours</i>	91.772	123.032	114.606	114.016	114.016	119.606	121.811
<b>Loss per year(mils)</b>	<b>2.069</b>	<b>2.759</b>	<b>2.586</b>	<b>0.862</b>	<b>1.207</b>	<b>0.517</b>	<b>2.759</b>

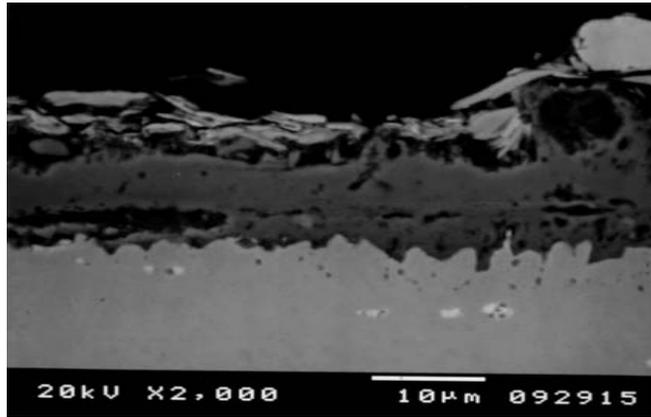
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\*Based on full time run (8760 hours/year)

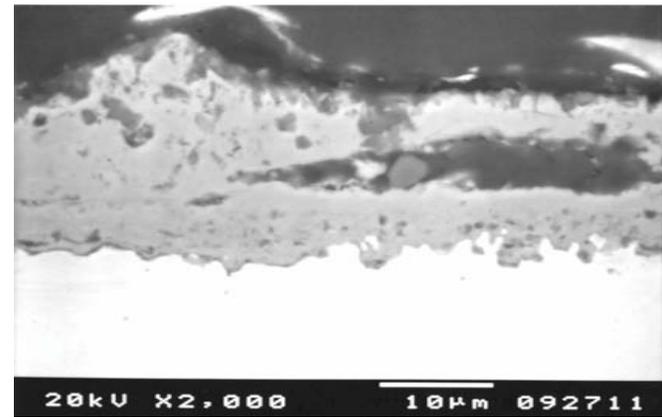
# Overview of Five Test Tubes



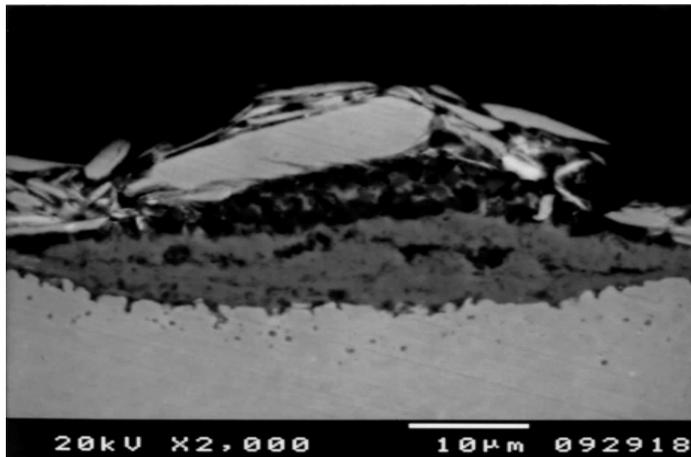
# Effect of Ash Deposits on Tubes: SEM Photos at Four Positions



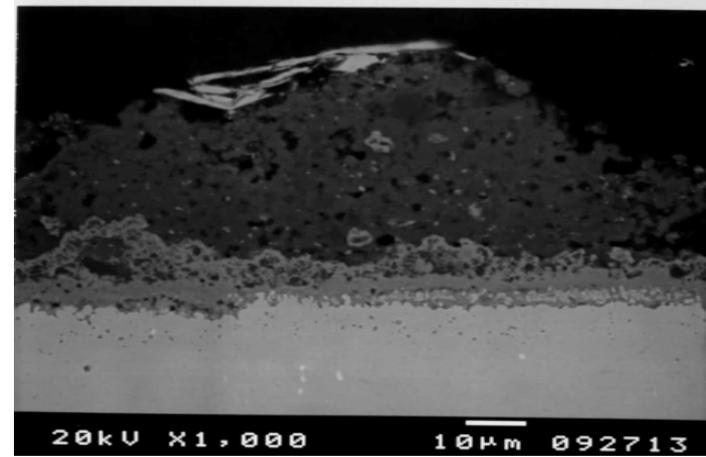
Surface of 12'O position



Surface of 6'O position

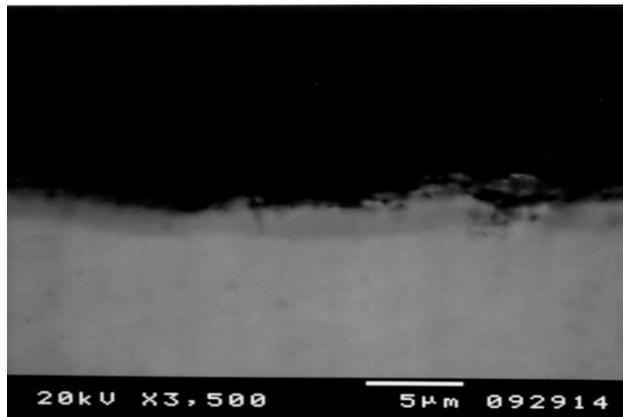


Outer edge of 12'O position

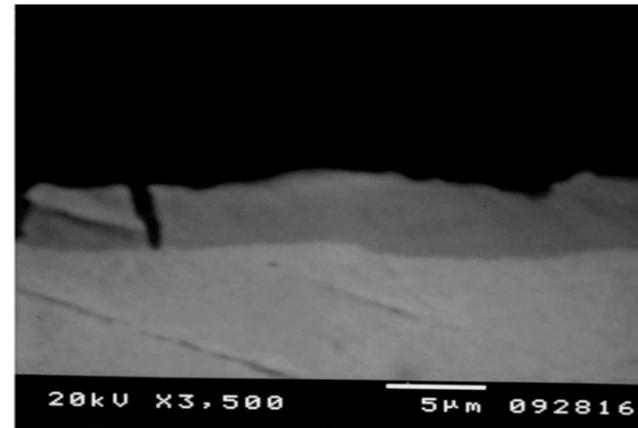


Outer edge of 6'O position

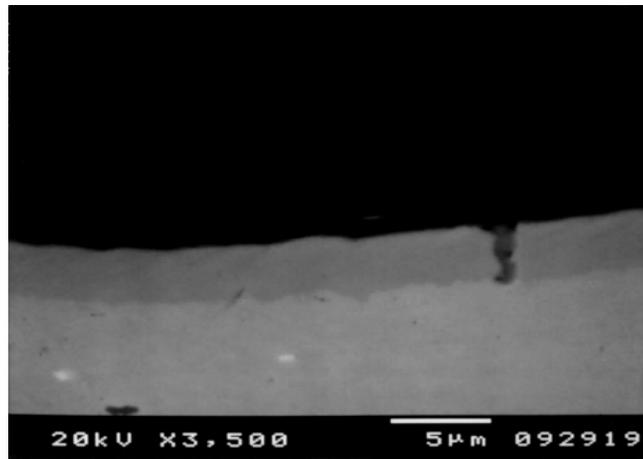
# Results from SEM-EDX Analysis of the Inner Edge of Coupons



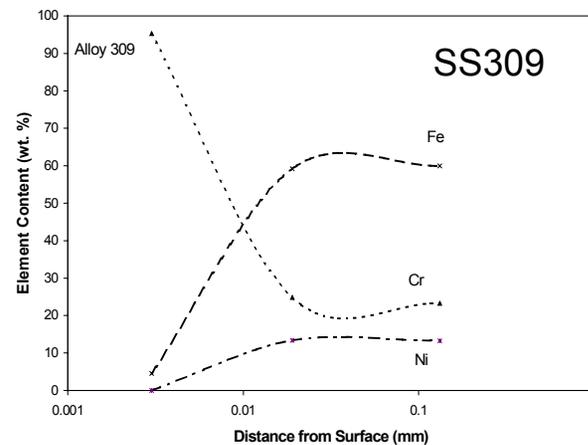
SS304



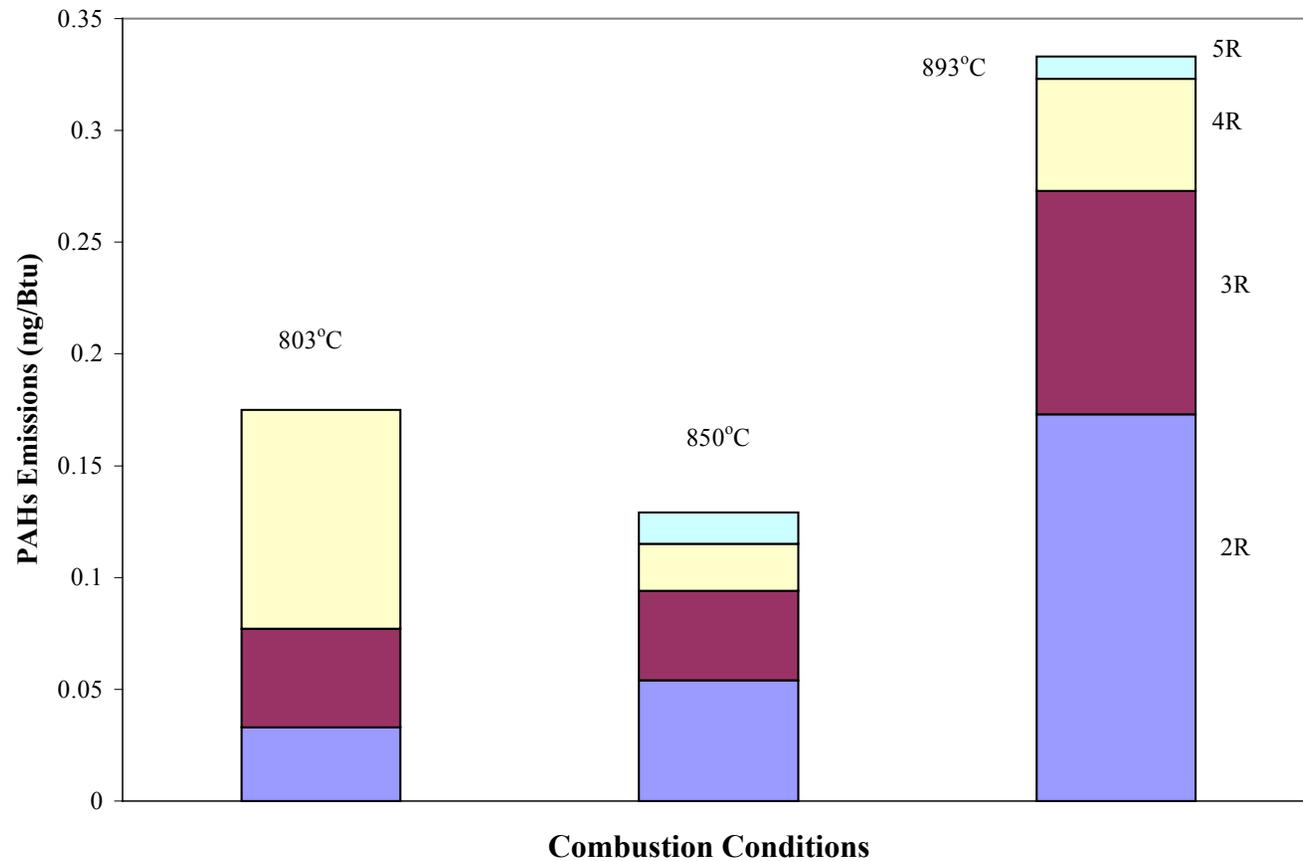
SS309



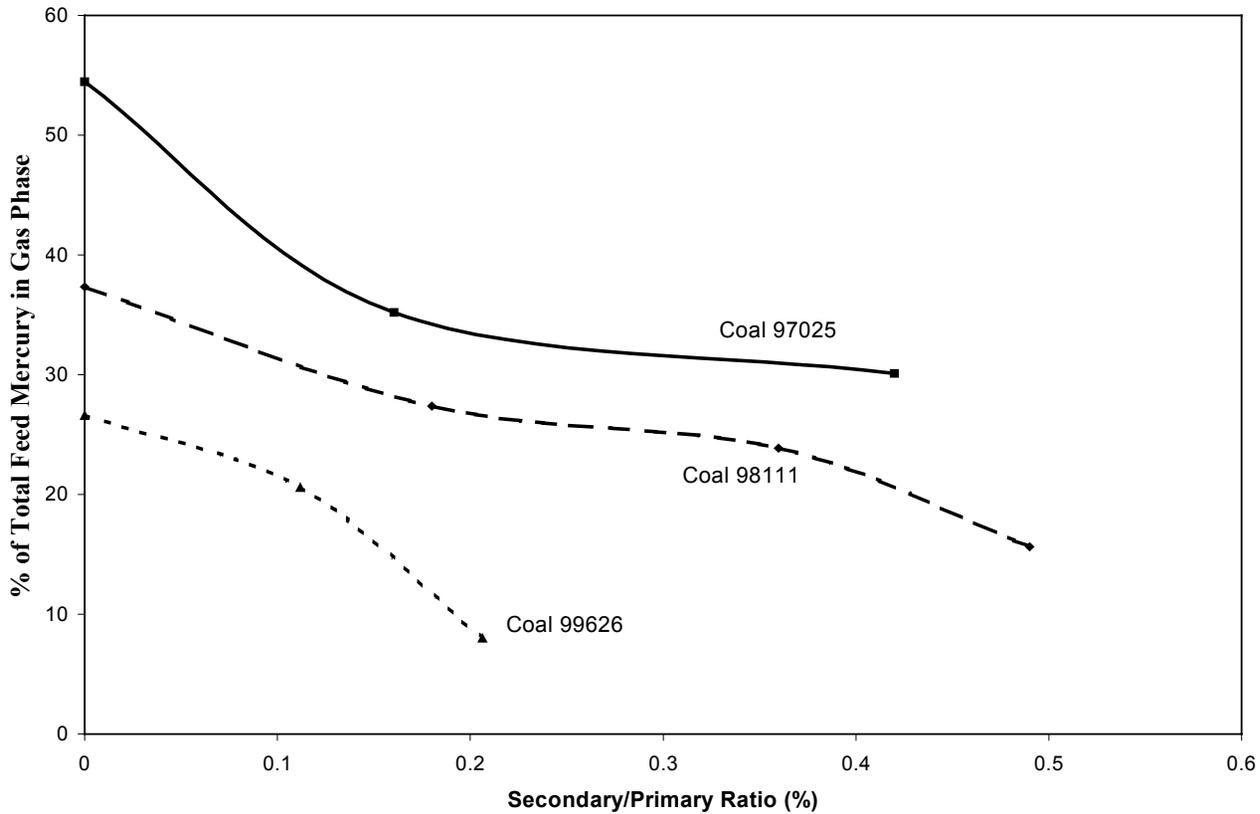
SS347



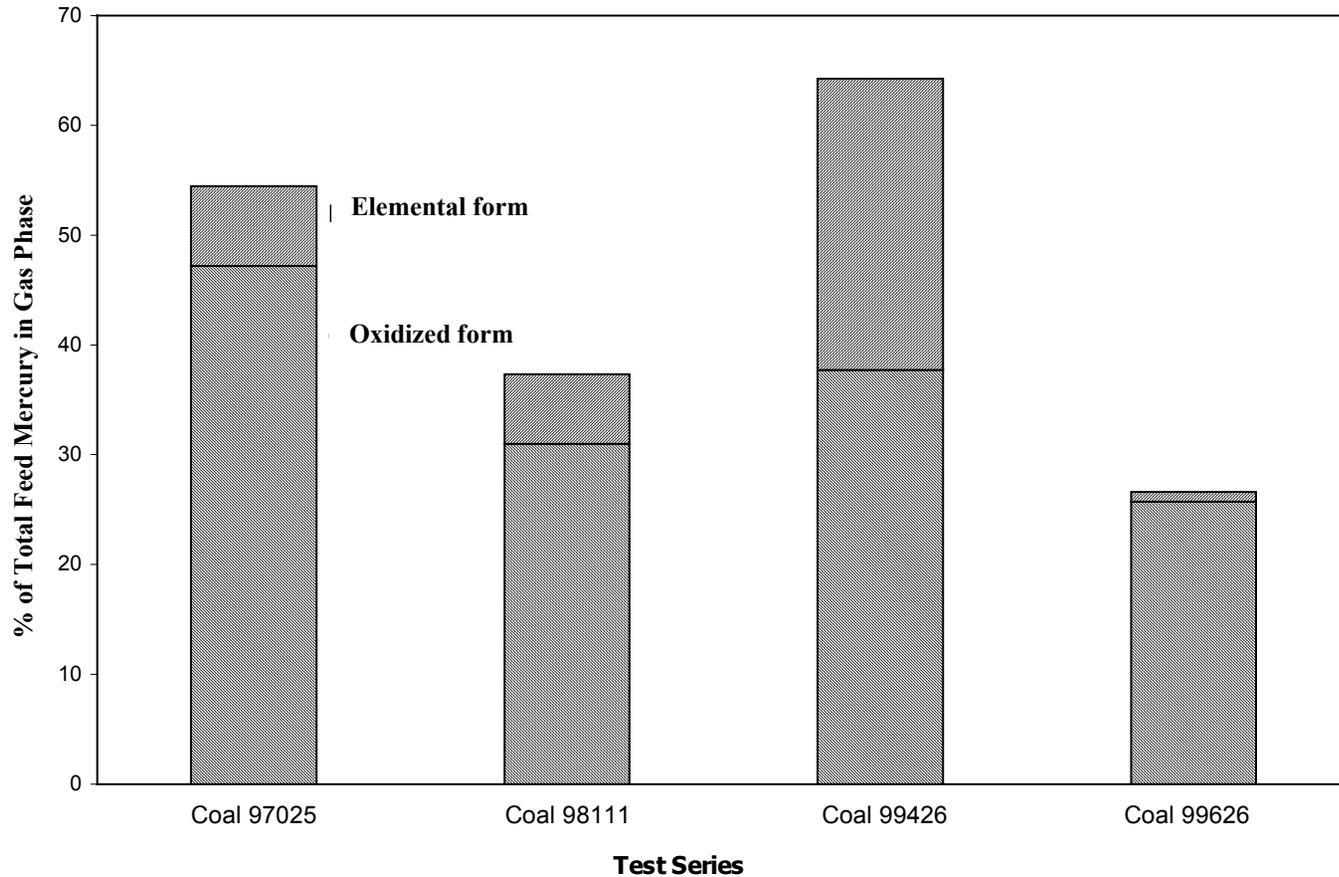
# Effect of Combustion Temperature on the Emissions of PAHs in the Flue Gas



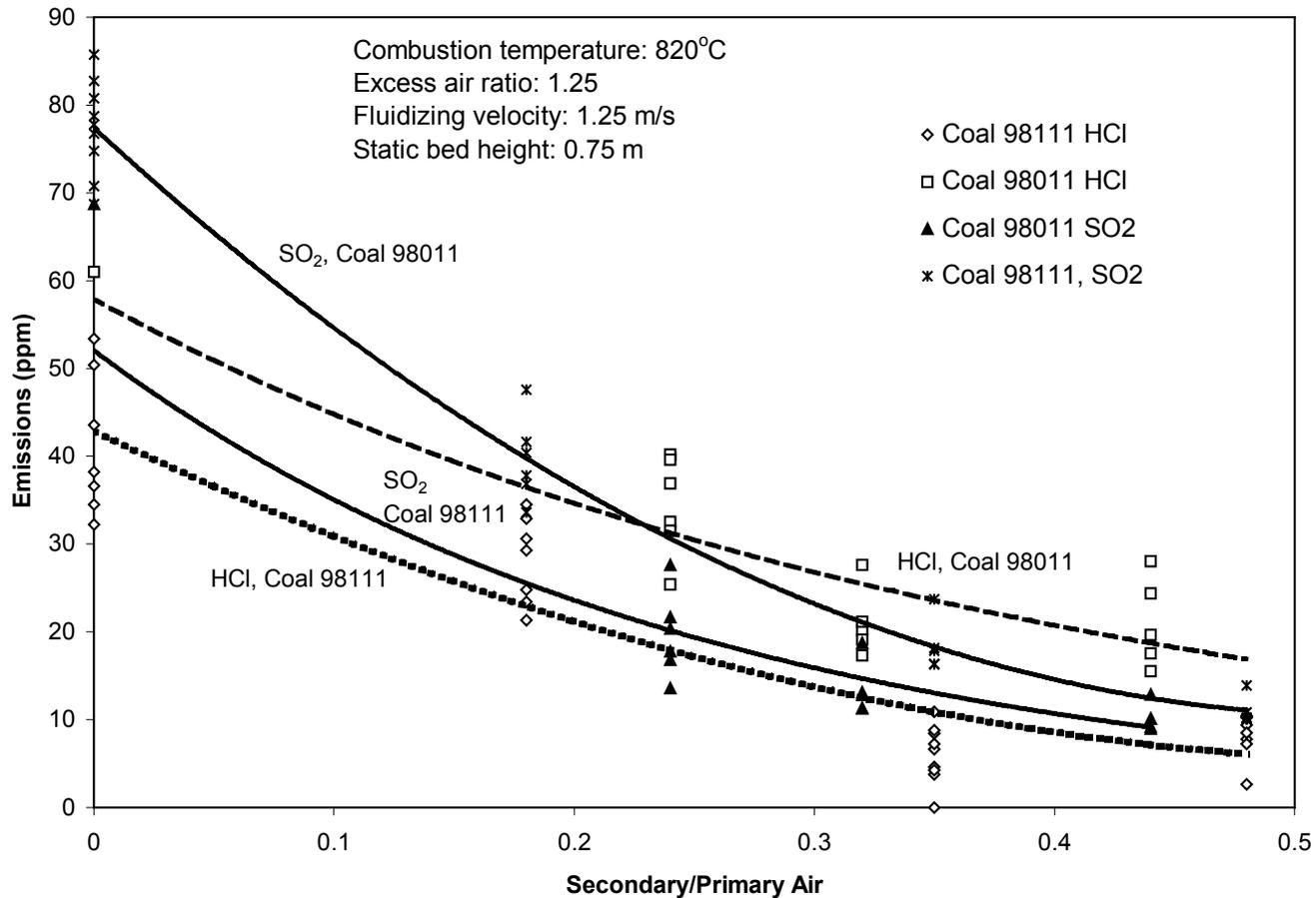
# Effect of Coal Types on the Emissions of Mercury in the Flue Gas



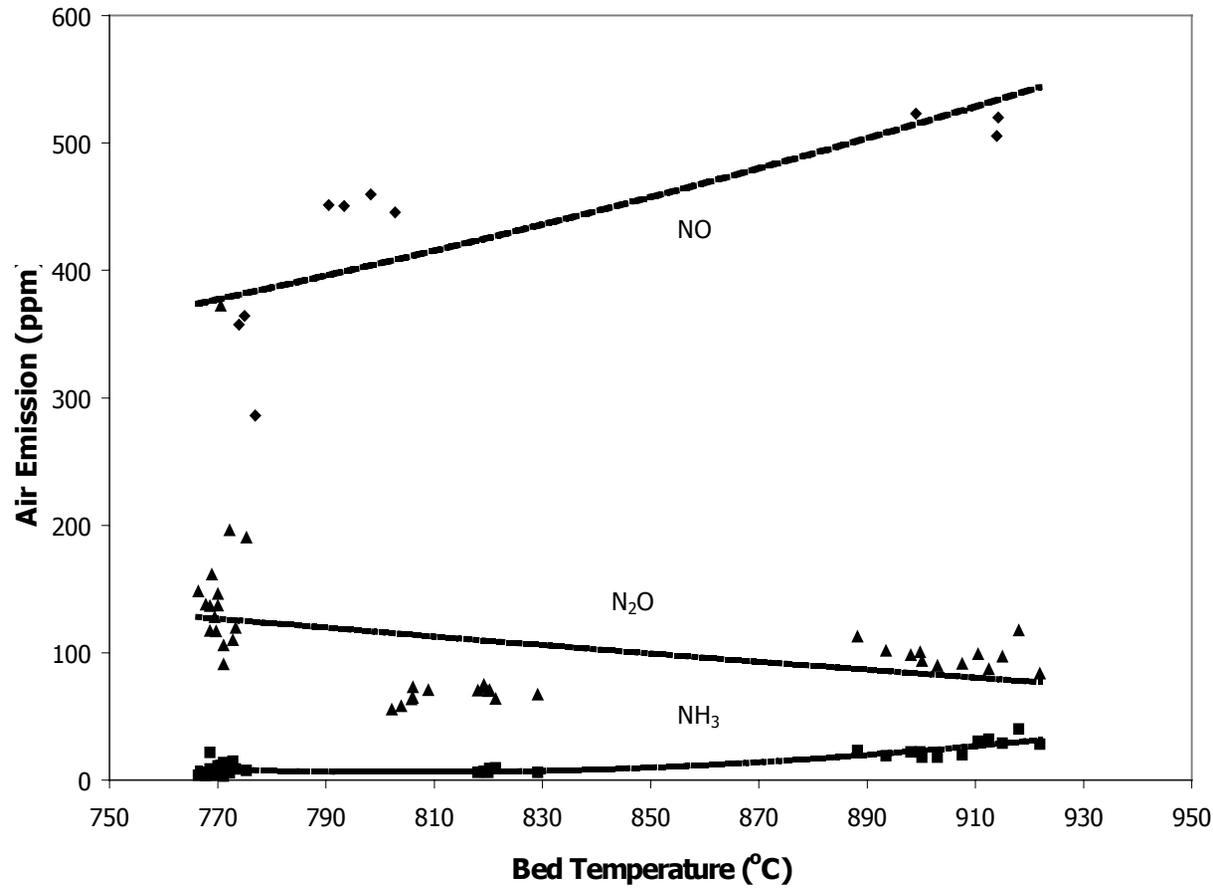
# Effect of Coal Types on Mercury Species Emission in Flue Gas



# Effect of Coal Types on the Emissions of HCl and SO<sub>x</sub> in the Flue Gas



# Effect of Combustion Temperature on Nitrogen Compound Emissions



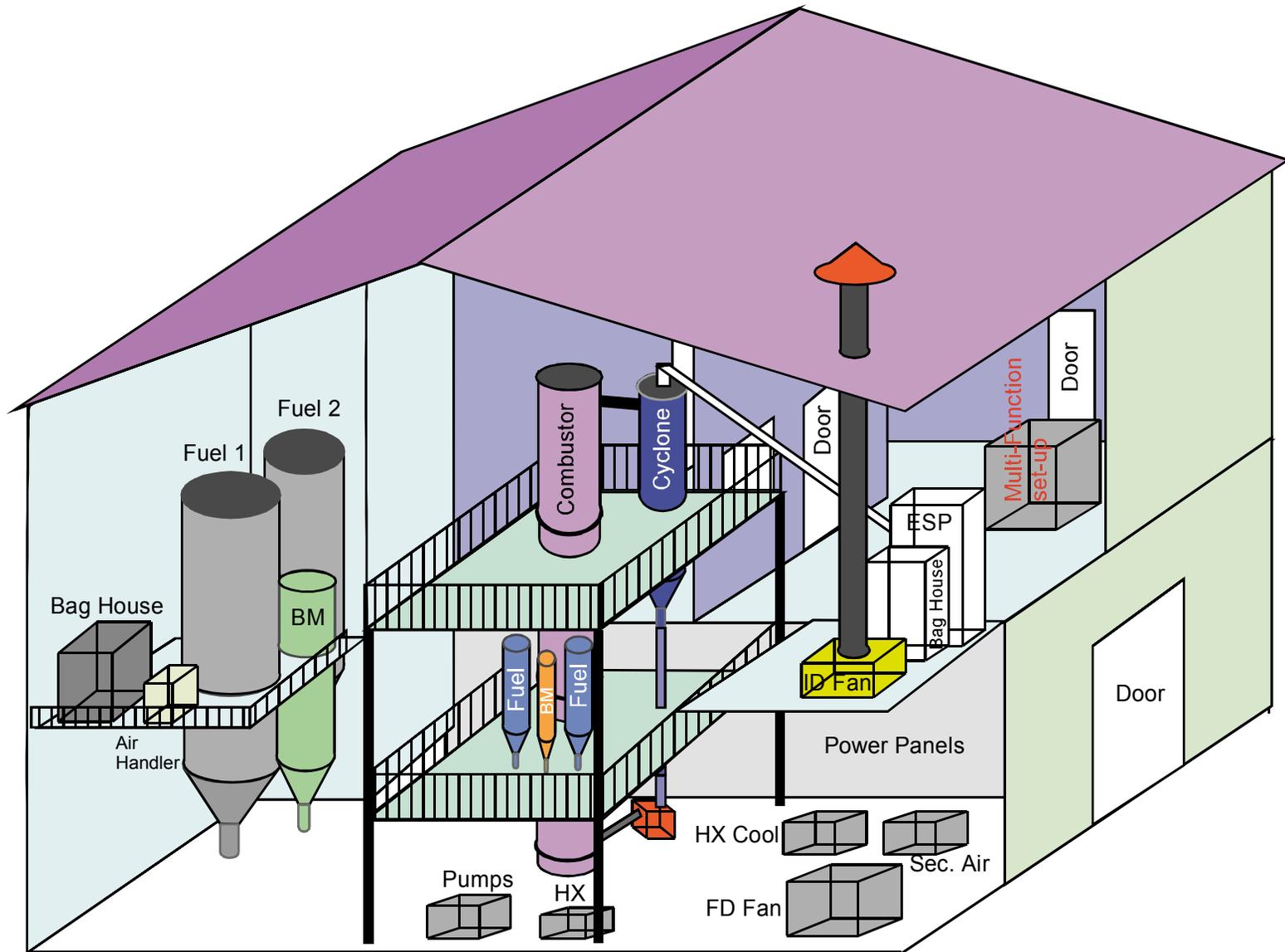
# Advantages of Co-firing MSW/RDF or Biomass with High Sulfur Coals

- Energy recovery
- Sulfur dioxide in combustion gases minimizes production of molecular chlorine to control the formation of polychlorinated organics.
- Convert elemental mercury to oxidized mercury in the presence of HCl and fly ash acting as catalysts in the upper combustion area of a utility boiler
- Reduce CO<sub>2</sub> emission.

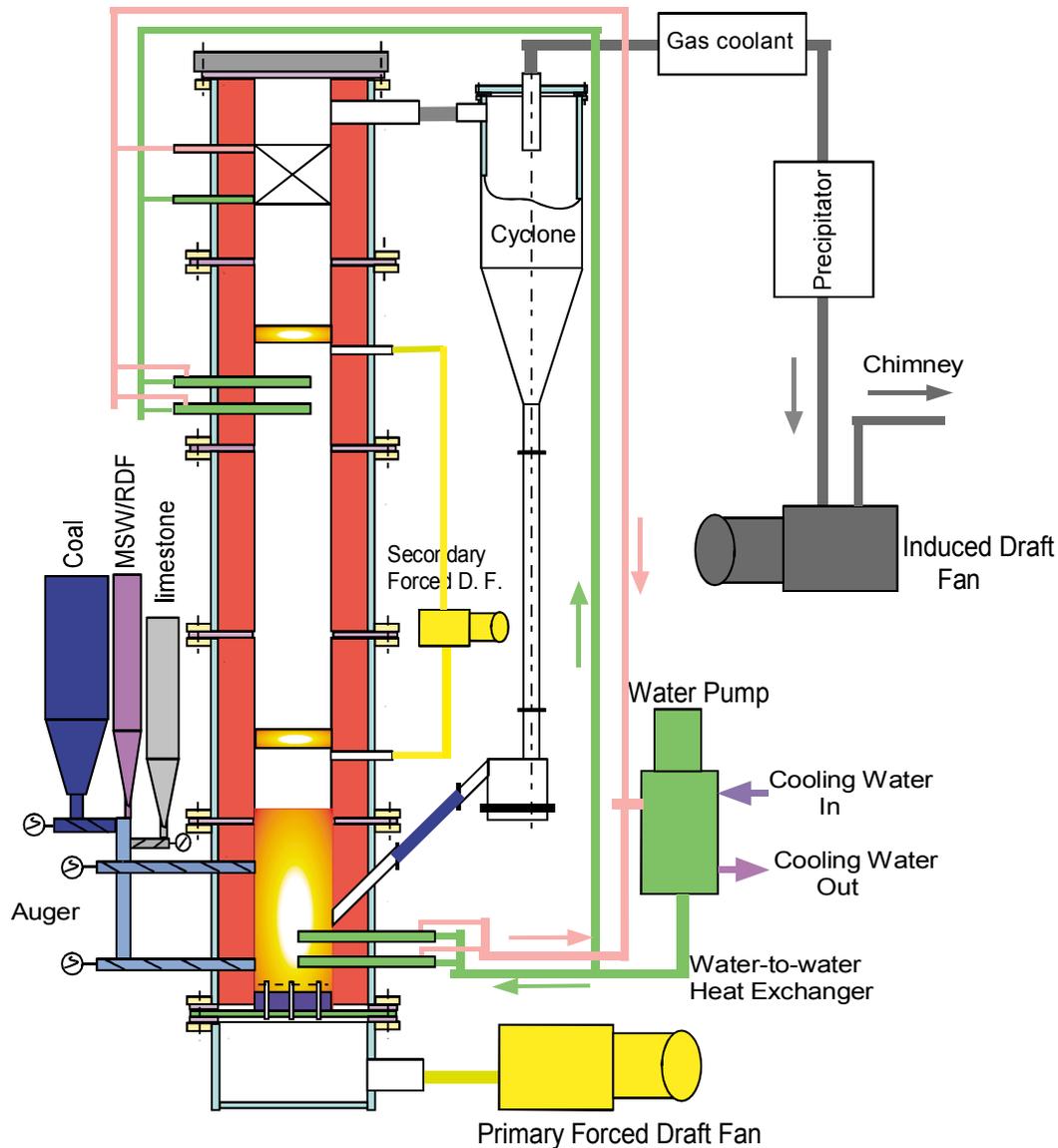
# Benefits

- Reduction in landfill costs, in carbon dioxide emissions, in chlorinated organic and PAH compound emissions, and in heavy metal and other greenhouse gases emissions
- Energy recovery from waste
- Open new markets for processed MSW/RDF, biomass and Western Kentucky, Illinois, and Ohio high sulfur coals.

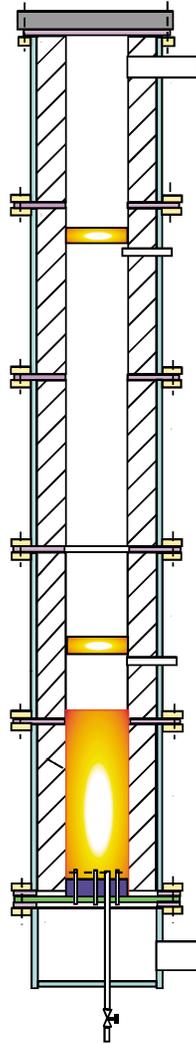
# 3D views of CFBC Layout



# Diagram of 0.6 MWth CFBC System



# Brief Design Parameters of 0.6 MWth CFBC System



The major parameters:

If inside diameter is chosen as 0.3 meter, 16-20 meters effective height  
and fluidizing velocity: 5 m/s (maximum)  
and combustion temperature: 850 °C

and coal feeding rate: 45-55 kg/h (~0.6 MWe)

Then, the capacity of F.D.F.:

Flow rate: 20 lb/min (minimum) (15 lb/min request)  
Pressure: 2500 mmH<sub>2</sub>O (99 inch WC) (max.)

the capacity of I.D.F.:

If we choose water exchange tube for flue gas cooling  
Flowrate: 25 lb/min (minimum)  
Pressure: 500 mmH<sub>2</sub>O (20 inch WC) (max.)

If we use cold air to mix with hot gas for flue gas cooling  
Flowrate and Pressure will depend on the temperature at  
outlet of cyclone and the maximum temperature of  
baghouse.

the capacity of secondary F.D.F.:

Flowrate: 9 lb/min  
Pressure: 1000 mmH<sub>2</sub>O (40 inch WC) (max.)

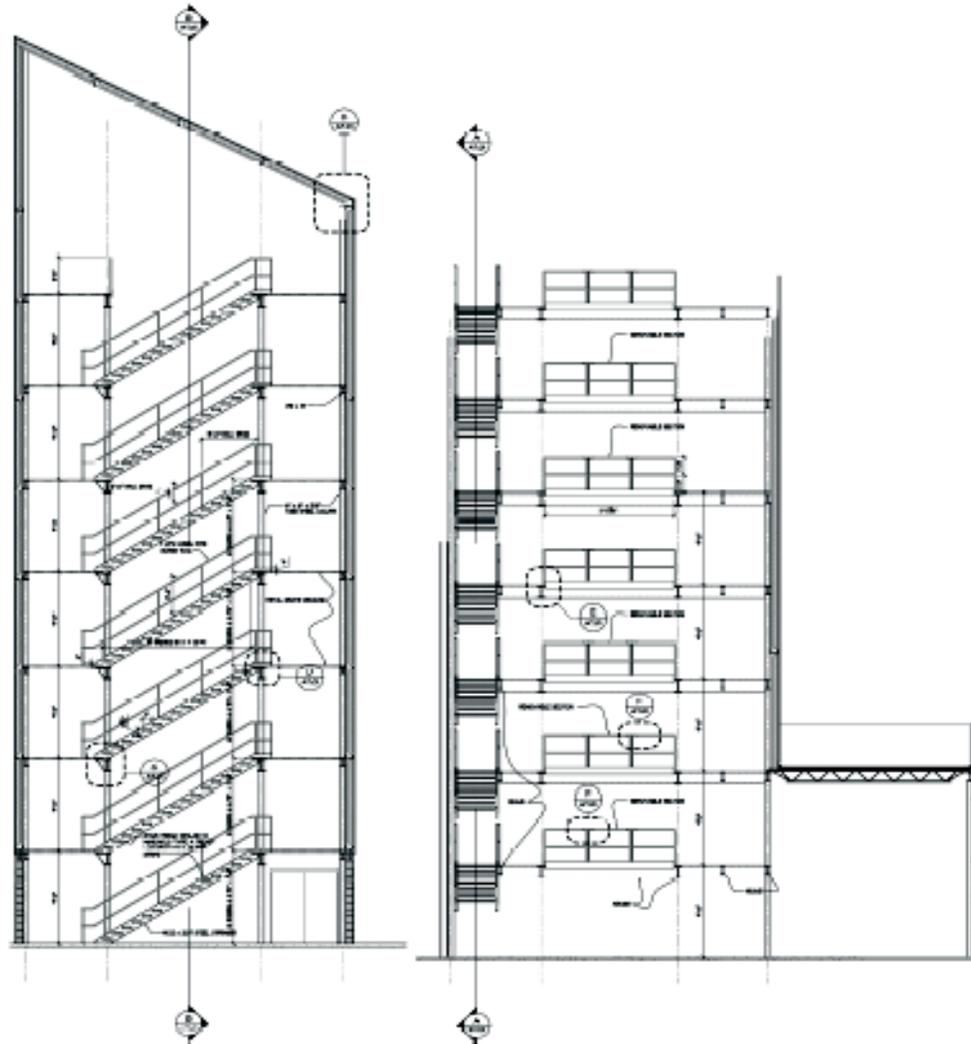
the set-plate:

Pressure drop: 150--250 mmH<sub>2</sub>O (6--10 inch WC)

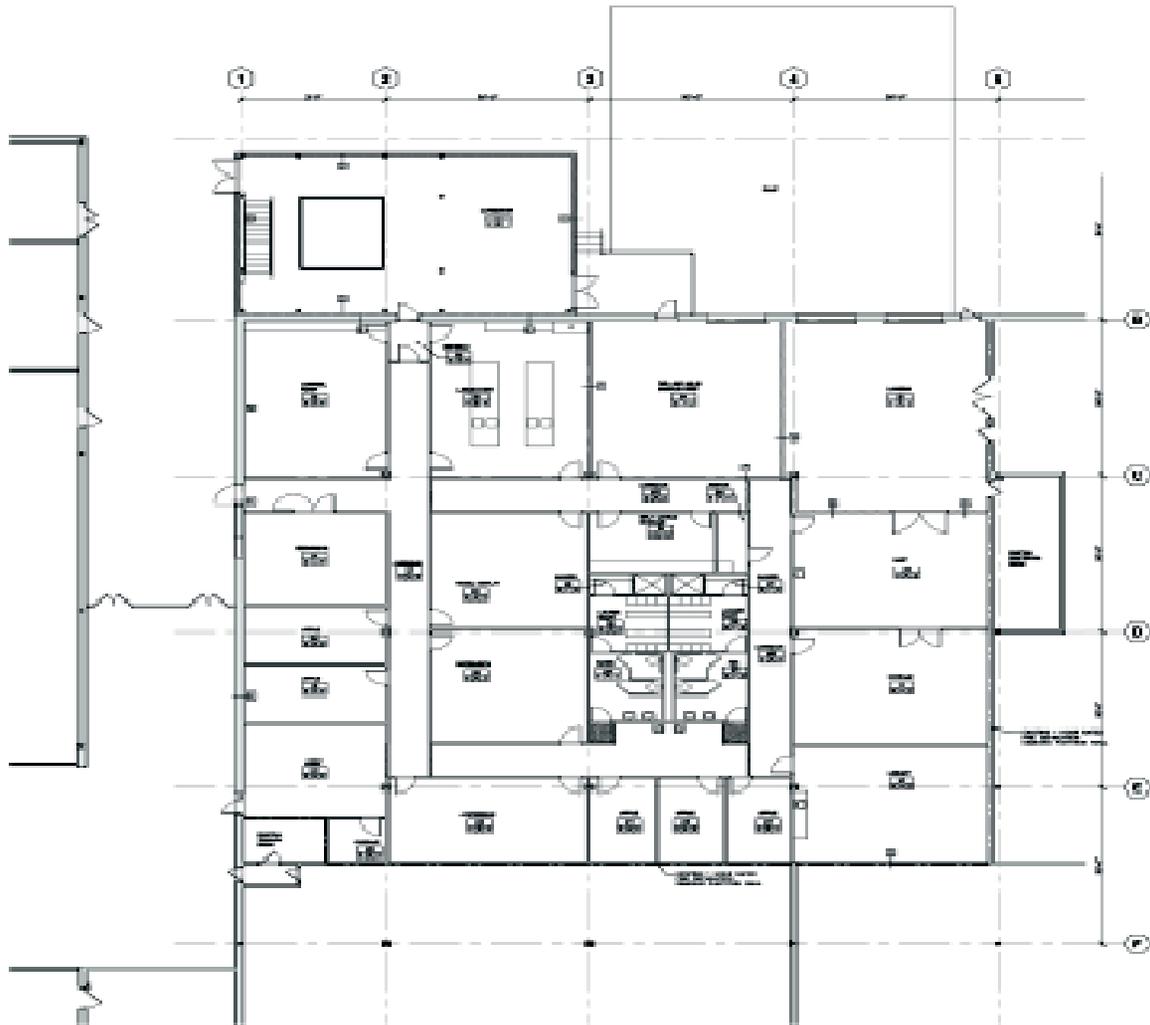
Estimated weight:

Combustor: 12,000 lbs  
Cyclone: 2,000 lbs  
Coal and limestone system: 2,000 lbs  
Heat exchange tubes: 1,000 lbs  
Platform: 2,000 lbs  
Another accessories: 3,000 lbs

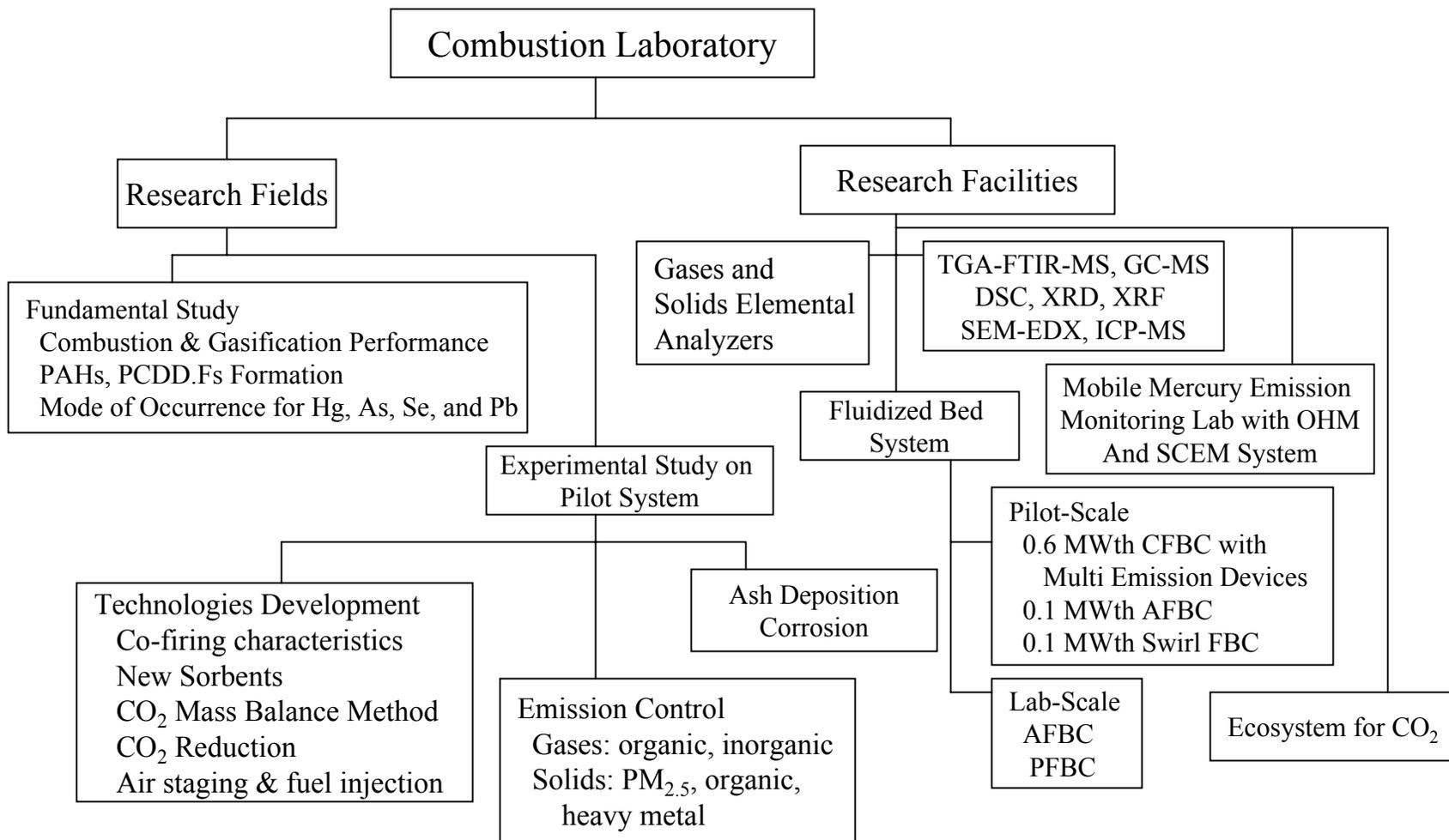
# Building for CFBC



# Combustion Laboratory and CFBC Facility

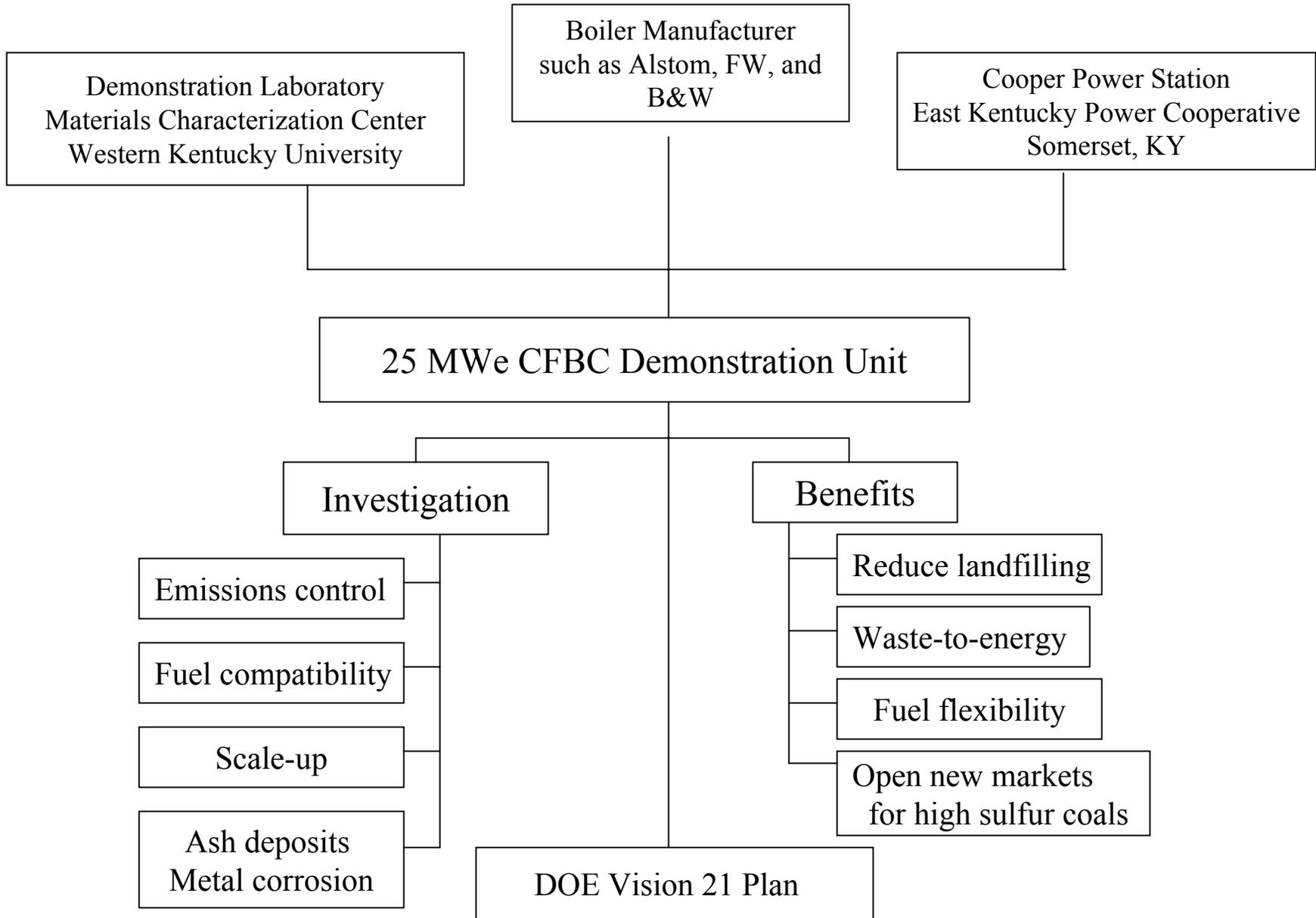


# Establishment of a Demonstration Laboratory for Co-firing Refuse Derived Fuels with High Sulfur Coals



# Future Work

- Transfer Co-firing RDF with high sulfur coal technology to a utility boiler.
- Use new sorbents to retrofit a boiler to capture  $\text{SO}_x$ , HCl and Hg with low operating cost.
- Study the performance of partial gasification of co-firing RDF with high sulfur coal in CFB and PFB systems.
- Build a relationship with Kentucky Pioneer Energy Project



Demonstration Laboratory  
Materials Characterization Center  
Western Kentucky University

Boiler Manufacturer  
such as Alstom, FW, and  
B&W

Cooper Power Station  
East Kentucky Power Cooperative  
Somerset, KY

25 MWe CFBC Demonstration Unit

Investigation

Emissions control

Fuel compatibility

Scale-up

Ash deposits  
Metal corrosion

Benefits

Reduce landfilling

Waste-to-energy

Fuel flexibility

Open new markets  
for high sulfur coals

DOE Vision 21 Plan